

AYDIN ADNAN MENDERES UNIVERSITY COURSE INFORMATION FORM

Course Title		Introduction to	Chemistry I							
Course Code		KMY161		Couse Level		Short Cycle (Associate's Degree)				
ECTS Credit 3		Workload	74 (Hours)) Theory		2	Practice	0	Laboratory	0
Objectives of the Course		improve students' ability to think about mater's properties and measurement, atoms and atomic theory, electronic structure of atoms, the periodic table and give theoretical knowledge in a systematic and comprehensive on some atomic properties and the basic concepts of chemistry								
Course Content		periodic prope	erties, electror npounds, read	nic struc ctions ar	ture of nd stoid	atoms, at chiometric	tomic mass an	d mole conce chemical bond	er, the periodic ta ot, chemical form ds, molecules an and bases	nulas,
Work Placement		N/A								
Planned Learning Activities and Teaching Methods			Explan	nation (Presenta	tion), Discussio	on, Problem S	olving		
Name of Lecturer(s)		Lec. Ali ERKU	L							

Assessment Methods and Criteria

Method	Quantity	Percentage (%)	
Midterm Examination	1	40	
Final Examination	1	70	

Recommended or Required Reading

- 1 Genel Kimya. Sabri Alpaydın Abdullah Şimşek Nobel Yayın Dağıtım, 2009
- 2 Öğretim üyesi ders notları.

Week	Weekly Detailed Co	ourse Contents					
1	Theoretical	The basic unit and unit systems Chemistry					
2	Theoretical	The classification and properties of the substance					
3	Theoretical	Periodic table and periodic properties					
4	Theoretical	electronic structure of atoms, atomic mass and mole concept					
5	Theoretical	Chemical formulas					
6	Theoretical	Nomenclature of Compound					
7	Theoretical	Reactions and stoichiometric calculations					
8	Theoretical	Chemical bonds					
9	Theoretical	Molecules and their properties					
10	Theoretical	Midterm					
11	Theoretical	Gases and solids					
12	Theoretical	Liquids and Solutions					
13	Theoretical	Solution calculations					
14	Theoretical	Acids and bases					
15	Theoretical	Final exam					

Workload Calculation

Activity	Quantity	Preparation	Duration	Total Workload
Lecture - Theory	14	0	2	28
Midterm Examination	1	22	1	23



					Course Information For
Final Examination	1		22	1	23
Total Workload (Hours)					
[Total Workload (Hours) / 25*] = ECTS					
*25 hour workload is accepted as 1 ECTS					

Learning Outcomes

1	To understand the aim of chemistry, material properties and the classification
2	To understand the first discoveries in chemistry, atomic theory and the structure of atoms
3	To understand the periodic table, and the number of moles Avogadro
4	To distinguish the periodic properties of elements, understand the types of chemical compounds, to make chemical formulas
5	being able to make stoichiometric calculations using chemical reactions and chemical reactions to distinguish equality
6	be able to understand covalent bonding, molecular geometry and hybridization of atomic orbitals

Programme Outcomes (Medical and Aromatic Plants)

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1	Understands the importance of medicinal and aromatic plants in the World and Turkey				
2	Learn about the general characteristics of medicinal and aromatic plants. Learn the important issues in cultivation and can apply.				
3	Learn about usage technologies about medicinal and aromatic plants and can apply.				
4	Inform of producers of medicinal and aromatic plant species in offering, material supply, production process, marketing matter.				
5	Know and follow the laws and regulations pertaining to the profession.				
6	Learns morphological and anatomical structures of plants.				
7	Learns to identify medicinal and aromatic plants.				
8	To be able to behave sensitively towards environmental issues at national and global levels and to be able to interpret solution-oriented information; to be able to be an environmentally conscious and entrepreneurial individual				
9	To be able to follow, evaluate and implement new developments and applications in the cultivation of medicinal and aromatic plants independently or as a team.				

Contribution of Learning Outcomes to Programme Outcomes 1: Very Low, 2: Low, 3: Medium, 4: High, 5: Very High

	L1	L2	L5
P2	4	4	
P3	1	1	5
P8			4
P9	4	5	4