



## AYDIN ADNAN MENDERES UNIVERSITY COURSE INFORMATION FORM

|  |   |   |                      |   |   |                                |   |            |   |
|--|---|---|----------------------|---|---|--------------------------------|---|------------|---|
| Course Title                                     |   | Transition Metal Chemistry  |                      |   |   |                                |   |            |   |
| Course Code                                      |   | KİM631  |                      | Couese Level  |   | Third Cycle (Doctorate Degree) |   |            |   |
| ECTS Credit                                      | 8 | Workload  | 204 ( <i>Hours</i> ) | Theory  | 3 | Practice                       | 0 | Laboratory | 0 |
| Objectives of the Course                         |   | The aim of this course is to give basic information about properties, nature, methods, reactions, compounds and usage areas of industry.  |                      |   |   |                                |   |            |   |
| Course Content                                   |   | Physical properties of metals, theories related to metal binding (free electron theory, valence bond theory, molecular orbital theory), conductors, semiconductors and IB-VIIIB group metals, properties, natural occurrences, methods of obtaining, reactions, compounds with other elements properties and usage areas. |                      |   |   |                                |   |            |   |
| Work Placement                                   |   | N/A   |                      |   |   |                                |   |            |   |
| Planned Learning Activities and Teaching Methods |   |   |                      | Explanation (Presentation), Discussion, Problem Solving |   |                                |   |            |   |
| Name of Lecturer(s)                              |   |   |                      |   |   |                                |   |            |   |

### Assessment Methods and Criteria

| Method              | Quantity | Percentage (%) |
|---------------------|----------|----------------|
| Midterm Examination | 1        | 40             |
| Final Examination   | 1        | 60             |

### Recommended or Required Reading

|   |  |
|---|--|
| 1 | Malcolm Gerloch, Edwin, C. Constable, Transition Metal Chemistry.  |
| 2 | P. W. Atkins, Inorganic Chemistry, D.F. Shriver; Oxford, 1999      |
| 3 | Tezcan, H., Tezcan, R. Metaller Kimyası Nobel Yayın Dağıtım, 2007. |

| Week | Weekly Detailed Course Contents |  |
|------|---------------------------------|--|
| 1    | Theoretical                     |  |
| 2    | Theoretical                     |  |
| 3    | Theoretical                     |  |
| 4    | Theoretical                     |  |
| 5    | Theoretical                     |  |
| 6    | Theoretical                     |  |
| 7    | Theoretical                     |  |
| 8    | Theoretical                     |  |
| 9    | Intermediate Exam               |  |
| 10   | Theoretical                     |  |
| 11   | Theoretical                     |  |
| 12   | Theoretical                     |  |
| 13   | Theoretical                     |  |
| 14   | Theoretical                     |  |
| 15   | Theoretical                     |  |
| 16   | Final Exam                      |  |

### Workload Calculation

| Activity            | Quantity | Preparation | Duration | Total Workload |
|---------------------|----------|-------------|----------|----------------|
| Lecture - Theory    | 14       | 0           | 3        | 42             |
| Assignment          | 4        | 20          | 0        | 80             |
| Reading             | 14       | 2           | 0        | 28             |
| Midterm Examination | 1        | 25          | 2        | 27             |



|   |   |    |   |     |
|---|---|----|---|-----|
| Final Examination                       | 1 | 25 | 2 | 27  |
| Total Workload (Hours)                  |   |    |   | 204 |
| [Total Workload (Hours) / 25*] = ECTS   |   |    |   | 8   |
| *25 hour workload is accepted as 1 ECTS |   |    |   |     |

### Learning Outcomes

|   |  |
|---|--|
| 1 |  |
| 2 |  |
| 3 |  |
| 4 |  |
| 5 |  |

### Programme Outcomes (Chemistry Doctorate)

|   |   |
|---|---|
| 1 | Depending on the master degree competences, develops, insights and innovates current and advanced knowledge and/or research in proficiency level.   |
| 2 | Gains high skill levels in using research methods in the field of his/her study.  |
| 3 | Comprehends the interaction between disciplines related to his/her field. Reaches to original results using his/her expertise in order to analyze, synthesize and evaluate new and complicated ideas. |
| 4 | Enlarges the boundaries of his/her field of knowledge by publishing at least one research paper in national and/or international peer-reviewed journals.  |
| 5 | Defends his/her original opinions related to his/her field before authority and communicates effectively illustrating his/her competence.   |
| 6 | May communicate and debate written, orally and visually in European Language Portfolio level C1.  |
| 7 | Follows the developments in computer software and information and communication technologies developed for his/her research area and uses these in order to solve research problems.                  |
| 8 | Collaborates for scientific research with national and international research teams.  |
| 9 | Contributes to the course of creation and maintenance of knowledge based society and by introducing the scientific, social and cultural developments to the society he/she is living in.              |

### Contribution of Learning Outcomes to Programme Outcomes 1:Very Low, 2:Low, 3:Medium, 4:High, 5:Very High

|    | L1 | L2 | L3 | L4 | L5 |
|----|----|----|----|----|----|
| P1 | 4  | 5  | 4  | 5  | 5  |
| P2 | 4  | 5  | 4  | 5  | 5  |

