



AYDIN ADNAN MENDERES UNIVERSITY COURSE INFORMATION FORM

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|--|---|--|----------------------|---|---|--------------------------------|---|------------|---|
| Course Title | | Quantum Mechanics | | | | | | | |
| Course Code | | FZK512 | | Couse Level | | Second Cycle (Master's Degree) | | | |
| ECTS Credit | 6 | Workload | 150 (<i>Hours</i>) | Theory | 3 | Practice | 0 | Laboratory | 0 |
| Objectives of the Course | | To introduce the physical structure of microsystems using quantum mechanics | | | | | | | |
| Course Content | | Angular momentum, addition of angular momentum, spherical symmetric potentials and their applications, indistinguishable particles, Helium atom, Hund rules, perturbation theory | | | | | | | |
| Work Placement | | | | | | | | | |
| Planned Learning Activities and Teaching Methods | | | | Explanation (Presentation), Individual Study, Problem Solving | | | | | |
| Name of Lecturer(s) | | | | | | | | | |

Assessment Methods and Criteria

| Method | Quantity | Percentage (%) |
|---------------------|----------|----------------|
| Midterm Examination | 1 | 20 |
| Final Examination | 1 | 30 |
| Quiz | 2 | 8 |
| Attending Lectures | 14 | 28 |
| Assignment | 14 | 14 |

Recommended or Required Reading

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| 1 | Introduction to Quantum Mechanics R. Shankar 2.Ed |
| 2 | Quantum Mechanics, Eugen Merzbacher, 3. Ed |
| 3 | Quantum Mechanics, Arno Bohm, 3. Ed. |
| 4 | Introduction to Quantum Mechanics, D.J. Griffiths |
| 5 | Kuantum Mekaniği Temel Kavramlar ve Uygulamaları, A. Verçin, T. Dereli |

| Week | Weekly Detailed Course Contents | |
|------|---------------------------------|--|
| 1 | Theoretical | Orbital angular momentum |
| | Preparation Work | Shankar R, Introduction to Quantum Mechanics, Chapter 11 |
| 2 | Theoretical | Algebraic approach to the angular momentum |
| | Preparation Work | Merzbacher E, Quantum Mechanics, Section 12.2, 12.3 |
| 3 | Preparation Work | Shankar R, Introduction to Quantum Mechanics, Chapter 14 |
| 4 | Theoretical | Addition of angular momentum |
| | Preparation Work | Merzbacher E, Quantum Mechanics, Section 17.4, 17.5 |
| 5 | Theoretical | Wigner-Eckart theorem |
| | Preparation Work | Bohm A, Quantum Mechanics, Section 5.3 |
| 6 | Theoretical | Spherically symmetric potentials |
| | Preparation Work | Shankar R, Introduction to Quantum Mechanics, Section 13.1-13.3 |
| 7 | Theoretical | Atoms with a single electron |
| | Preparation Work | Shankar R, Introduction to Quantum Mechanics, Section 13.4, 13.5 |
| 8 | Intermediate Exam | Midterm Exam |
| 9 | Theoretical | Indistinguishable particles |
| | Preparation Work | Dereli T. Verçin A, Kuantum Mekaniği, Section 10.1-10.4 |
| 10 | Theoretical | An elementary introduction to quantum statistics |
| | Preparation Work | Griffiths D.J., Introduction to Quantum Mechanics, Section 5.4, |
| 11 | Theoretical | Helium atom and Hund rules |
| | Preparation Work | Dereli T. Verçin A, Kuantum Mekaniği Section, 10.5-10.6 |
| 12 | Theoretical | Time independent perturbation theory |
| | Preparation Work | Shankar R, Introduction to Quantum Mechanics, Section 17.1, 17.2 |
| 13 | Theoretical | Time independent perturbation theory- Degenerate cases |
| | Preparation Work | Shankar R, Introduction to Quantum Mechanics, Section 17.3, |



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|----|------------------|---|
| 14 | Theoretical | Time dependent perturbation theory |
| | Preparation Work | Shankar R, Introduction to Quantum Mechanics, Section 18.1-18.3 |
| 15 | Theoretical | Photons and electromagnetic fields |
| | Preparation Work | Shankar R, Introduction to Quantum Mechanics, Section 18.4 |
| 16 | Final Exam | Final Exam |

Workload Calculation

| Activity | Quantity | Preparation | Duration | Total Workload |
|---|----------|-------------|----------|----------------|
| Lecture - Theory | 14 | 2 | 3 | 70 |
| Assignment | 14 | 2 | 0 | 28 |
| Quiz | 2 | 4 | 0.5 | 9 |
| Midterm Examination | 1 | 15 | 3 | 18 |
| Final Examination | 1 | 22 | 3 | 25 |
| Total Workload (Hours) | | | | 150 |
| [Total Workload (Hours) / 25*] = ECTS | | | | 6 |
| *25 hour workload is accepted as 1 ECTS | | | | |

Learning Outcomes

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|---|---|
| 1 | Student should be able to derive the algebra of angular momentum |
| 2 | Student should be able to use approximation methods in quantum mechanics |
| 3 | Student should be able to explain atomic structure using quantum mechanics |
| 4 | Student should know the quantum mechanics of the indistinguishable particles and discuss the effects to the statistical mechanics |
| 5 | Student should know the simple quantized theory of the electromagnetic fields based on the harmonic oscillator algebra |

Programme Outcomes (Physics Master)

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|---|--|
| 1 | The student should conceive the concepts in physics and may apply them on her/his own |
| 2 | The student should be able to conceive the relationship between the different physics laws and integrity of them and apply them in solving different physics problems |
| 3 | The student should know the basic principles of classical, quantum and relativistic physics and use them in the solutions of problems |
| 4 | The student should be able to do research in a specific area of physics |
| 5 | The student should be able to prepare reports on papers on the subject of her/his research and present her/his research subject in scientific conferences |
| 6 | The student should be able to explain the relationship between complicated problems and basic physics laws. |
| 7 | The student should be able to use computers for solving complicated physics problems |
| 8 | The student should be able to interrelate between the theory and the experiment. If she/he is experimentalist he/she has to explain the theory behind her/his work. If she /he is a theorist she/he should has to know the experiments in her/his subject. |

Contribution of Learning Outcomes to Programme Outcomes 1:Very Low, 2:Low, 3:Medium, 4:High, 5:Very High

| | L1 | L2 | L3 | L4 | L5 |
|----|----|----|----|----|----|
| P1 | 4 | 4 | 5 | 5 | 5 |
| P2 | 3 | 4 | 5 | 5 | 5 |
| P3 | 2 | 5 | 5 | 5 | 5 |
| P4 | 2 | 4 | 3 | 3 | 3 |
| P5 | 1 | 1 | 1 | 1 | 1 |
| P6 | 4 | 3 | 2 | 2 | 2 |
| P7 | 1 | 1 | 2 | 2 | 2 |
| P8 | 1 | 3 | 2 | 2 | 2 |

