

AYDIN ADNAN MENDERES UNIVERSITY COURSE INFORMATION FORM

Course Title General Chemistry										
Course Code	KMT109		Couse Level		Short Cycle (Associate's Degree)					
ECTS Credit 3	Workload 80 (Hours)		Theory	/	2	Practice	0	Laboratory	0	
and atomic theories, electron s					on structure of atom, periodic table and some atomic properties and to think about basic concepts of chemistry					
Course Content	periodic prope Nomenclature their properties	rties, Electror of compound s, Gases and lids, Liquids,	nic structis, Read solids, I	ture o tions Liquid	of atom, Ato and stoich is and Solu	omic mass and iometric calcul itions, Calcula	d mole concep lations, Chemi tions of Solutions	ter, Periodic table ot, Chemical form ical bonds, Molecons, Acids and B alculations of Sol	nulas, cules and sases,	
Work Placement N/A										
Planned Learning Activities and Teaching Methods			Explan	ation	(Presentat	tion), Individua	l Study, Probl	em Solving		
Name of Lecturer(s)										

Assessment Methods and Criteria						
Method	Quantity	Percentage (%)				
Midterm Examination	1	40				
Final Examination	1	70				

Recommended or Required Reading

1 Genel Kimya. Sabri Alpaydın - Abdullah Şimşek Nobel Yayın Dağıtım, 2009

Week	Weekly Detailed Course Contents							
1	Theoretical	Basic terms in chemistry and unit systems						
2	Theoretical	Classification and properties of matter						
3	Theoretical	Periodic table and periodic properties						
4	Theoretical	Electronic structure of atom, atomic masses and moles						
5	Theoretical	Chemical formulas, Chemical bonds						
6	Theoretical	Nomenclature of compounds						
7	Theoretical	Molecules and their properties						
8	Intermediate Exam	midterm						
9	Theoretical	Liquids and Solutions						
10	Theoretical	Solutions and numerical properties of solutions						
11	Theoretical	Solution calculations						
12	Theoretical	Solution calculations						
13	Theoretical	Acids and bases						
14	Theoretical	Acid-base equilibria						
15	Theoretical	Buffer solutions						
16	Final Exam	final exam						

Workload Calculation				
Activity	Quantity	Preparation	Duration	Total Workload
Lecture - Theory	14	0	2	28
Reading	14	0	2	28
Individual Work	6	0	2	12
Midterm Examination	1	5	1	6



Final Examination	1		5	1	6	
			To	tal Workload (Hours)	80	
			[Total Workload (Hours) / 25*] = ECTS	3	
*25 hour workload is accepted as 1 ECTS						

Learning Outcomes							
1	To understand the purpose of chemistry, properties and classification of matter						
2	To be able to comprehend the first discoveries in chemistry, atomic theory and atomic structure						
3	To be able to comprehend periodic table, mole and avogadro number						
4	Recognize solutions and make basic concentration calculations in solutions						
5	Recognize solutions and make basic concentration calculations in solutions						
6	Covalent bonding, comprehend atomic orbitals						
7	To be able to recognize acid, base and buffer solutions						

Programme Outcomes (Cosmetic Technology) To know the classification of cosmetic raw materials, for what purpose, in which products and how much they should be used 2 Define and classify cosmetics, To define, classify toxicity, Toxic substances and detoxification ways of these substances to know. To be able to analyze toxic 3 To be aware of the precautions to be taken when working with hazardous chemicals in terms of laboratory safety and human 4 health. 5 To have the ability to use basic mathematical methods to produce solutions. To be able to define the carrier systems used in cosmetics, to be able to choose the carrier system to be used according to the 6 cosmetic product. 7 To know and apply the necessary tests in cosmetic raw materials, intermediate products and finished products. Depending on the Atatürk nationalism in accordance with Atatürk's principles and reforms, adopted the national, moral, spiritual and cultural values of the Turkish Nation, and has adopted that the Turkish language is a rich, rooted and productive 8 language; have love and awareness of language; to have the ability to use the foreign language sufficiently to have the pleasure and habit of reading and need professionally.

Contribution of Learning Outcomes to Programme Outcomes 1:Very Low, 2:Low, 3:Medium, 4:High, 5:Very High									
	L1	L2	L3	L4	L5	L6	L7		
P1	3	4	4	3	2	2	3		
P2	4	4	4	4	3	2	3		
P3	5	3	4	3	2	3	4		
P4	3	4	5	3	2	2	4		
P5	4	3	4	4	2	1	5		
P6	5	3	4	2	2	2	5		
P7	5	4	3	2	5	4	4		

